

Jan 12-8:41 AM

Activity Series LAB

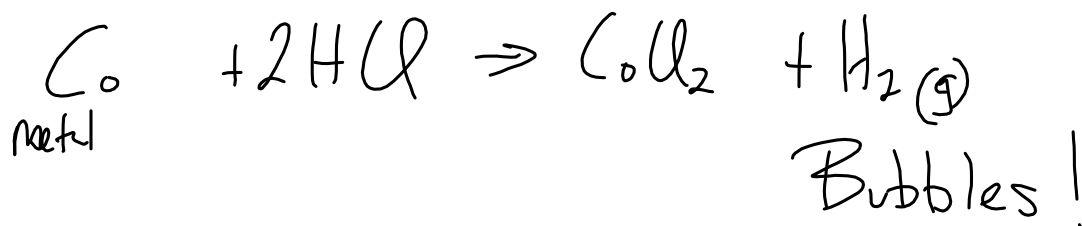
Table I  $\text{H}_2\text{SO}_4$

Zn	Fe	Zn Fe Mg Cu
Mg	Cu	

$\text{Cu}^{+2}$


$\text{Pb}^{+2}$

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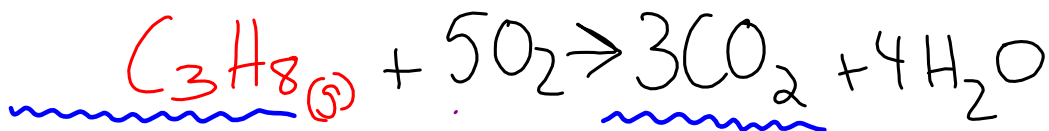
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Stoich

When in doubt ...

CONVERT TO MOLES!

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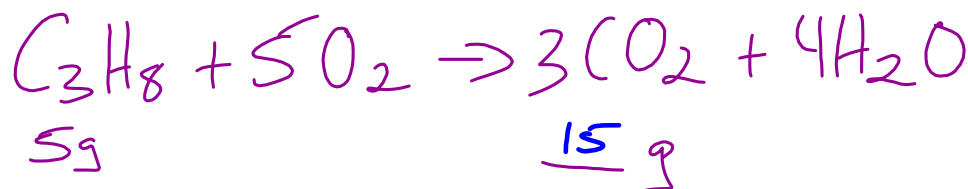


10L

3 CO<sub>2</sub> will be produced after combustion of 10L C<sub>3</sub>H<sub>8</sub>

10L C <sub>3</sub> H <sub>8</sub>	<del>1 mole C<sub>3</sub>H<sub>8</sub></del>	<del>3 mole CO<sub>2</sub></del>	<del>22.4L CO<sub>2</sub></del>	= 30L CO <sub>2</sub>
	<del>22.4L C<sub>3</sub>H<sub>8</sub></del>	1 mole C <sub>3</sub> H <sub>8</sub>	1 mole CO <sub>2</sub>	

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5g C <sub>3</sub> H <sub>8</sub>	<del>1 mole C<sub>3</sub>H<sub>8</sub></del>	<del>3 mole CO<sub>2</sub></del>	<del>44g CO<sub>2</sub></del>	=
	<del>44g C<sub>3</sub>H<sub>8</sub></del>	1 mole C <sub>3</sub> H <sub>8</sub>	1 mole CO <sub>2</sub>	

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