

$$2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$$

$$244\text{g}$$

244 g KClO₃	1 mole KClO₃	3 O₂	32g
23g KClO ₃ 122	2 KClO ₃	1 mole O ₂	

\rightarrow Mole_s

Mole Ratio
 \rightarrow Mole_s

\rightarrow 1
 \rightarrow 1 mole
 \rightarrow 3 molecules O₂

Jan 13-9:22 AM

Limiting Reactant (Limiting Reagent)

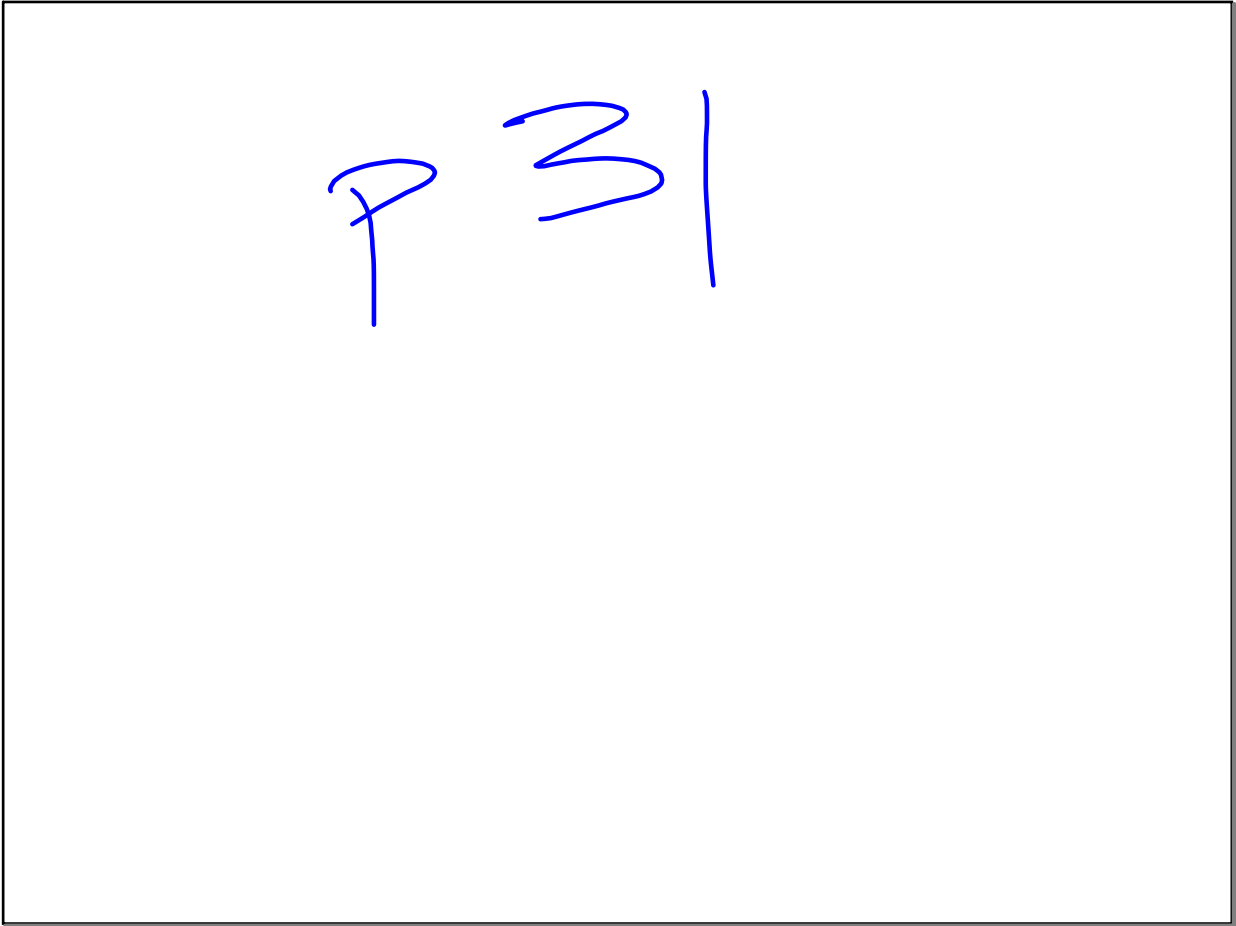
LR

1 Frames + 4 tires \rightarrow 1 CARs
 20 20 5?

20 Frames	1 car	= 20 cars
1 frame	1 frame	

20 Tires	1 cars	= 5 cars	* LR
4 Tires	4 Tires		

Jan 13-9:49 AM



Jan 13-10:02 AM