

$\overset{+1}{\boxed{\text{Cu}}} \overset{-1}{\boxed{\text{C}_2\text{H}_3\text{O}_2}}$

Copper (I) acetate

Copper (II) acetate

$\overset{+2}{\text{Cu}} \overset{-1}{\boxed{\text{C}_2\text{H}_3\text{O}_2}}$

$\text{Cu} \left(\text{C}_2\text{H}_3\text{O}_2 \right)_2$

Binary Compound

2 items make compound
 \oplus ox and \ominus ox

KClO₃

Nov 13-9:23 AM

~~$\overset{+3}{\text{Fe}_2} \overset{-2}{\text{O}_3}$~~

$\overset{+2}{\text{C}} \overset{-2}{\text{O}}$

Carbon Monoxide

Carbon (II) oxide

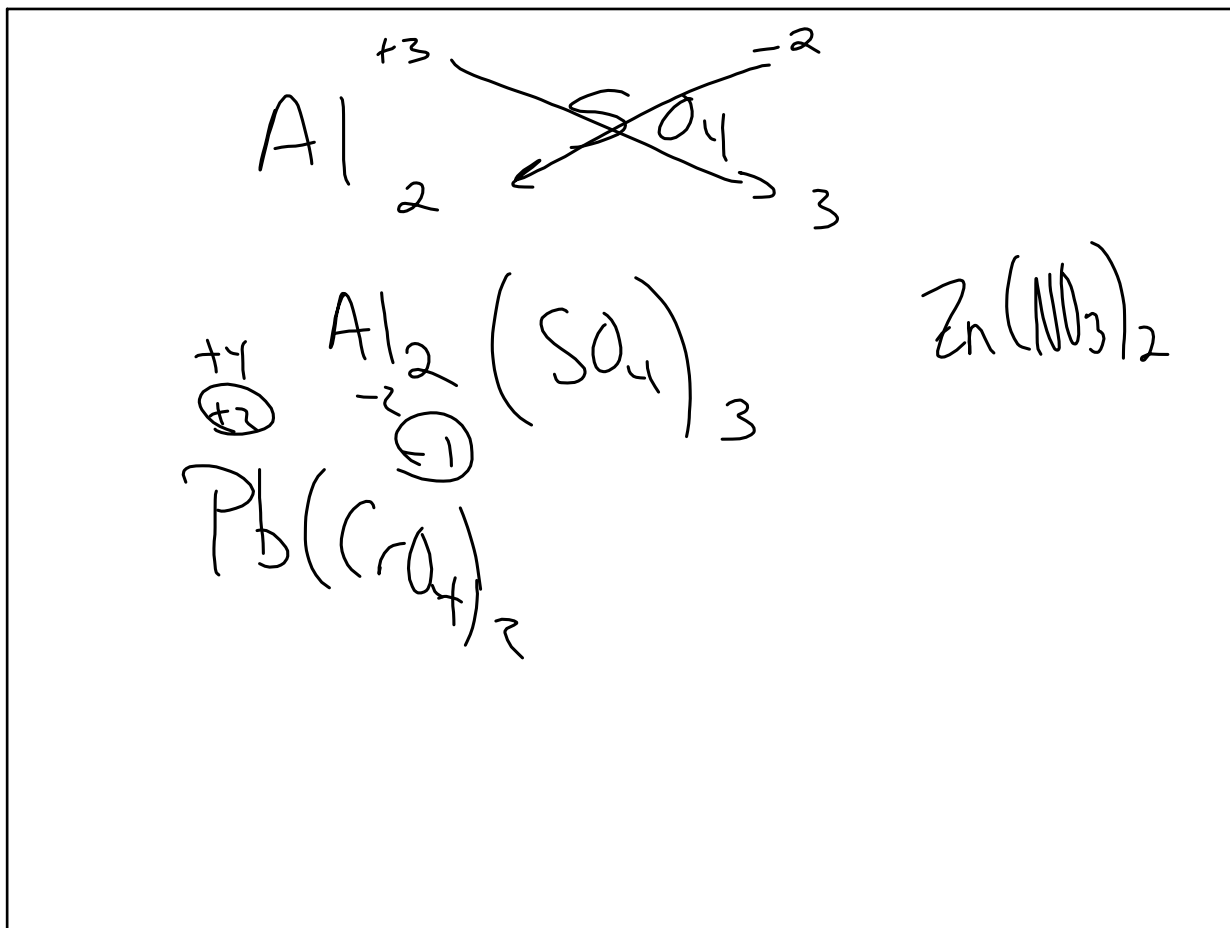
$\overset{+2}{\text{Fe}} \overset{-2}{\text{O}}$

$\overset{+4}{\text{C}} \overset{-2}{\text{O}_2}$

Carbon (IV) oxide

C₂O₄

Nov 13-9:44 AM



Nov 13-9:54 AM

Naming Acids

① H + element
 HCl
 hydro (Prefix) chloric acid
 chloric

② H + Polyatomic ion
 HNO_3
 Nitric acid
 HNO_2 → Nitrous acid
 $HClO_3$ → Chloric acid

③ ate - ic
ite - ous

Nov 13-9:57 AM

P 6+7 multiples of 3

Nov 13-10:03 AM