

Kool Aid - Lab write up.

Tomorrow

→ Show calc for Req strength  
then triple for 3X

→ Data / observation,

→ Discussion

Dec 17-9:23 AM

③ 0.425 g Lead Sulfate in 100g H<sub>2</sub>O

$$\text{PPM} = \frac{\text{Part}}{\text{Whole}} \times 1,000,000$$

Dec 17-9:57 AM

$M$  <sup>molarity</sup> = concentration  $\rightarrow$  "Strengths"  
of acids + Bases.

6M HCl  
Six Molar hydrochloric acid

Dec 17-9:59 AM

Make 6M HCl

6 moles HCl  
1 L

#9

6 moles HCl	36g	= 216g
1 mole		

Dec 17-10:02 AM

P465 #1 + 3

Dec 17-10:05 AM