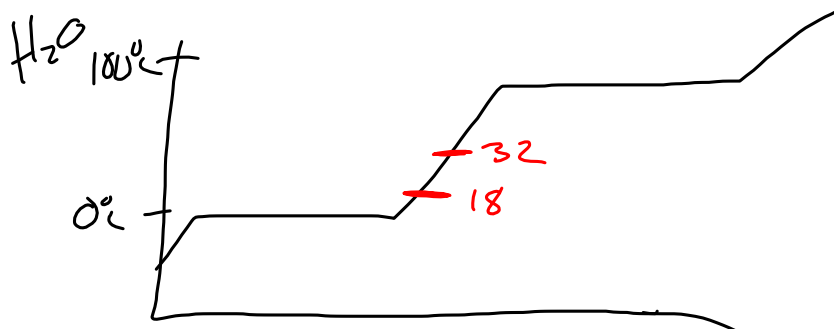


Feb 16-10:50 AM



$$Q = mc\Delta T$$

$$= (50)(4.18)(14)$$

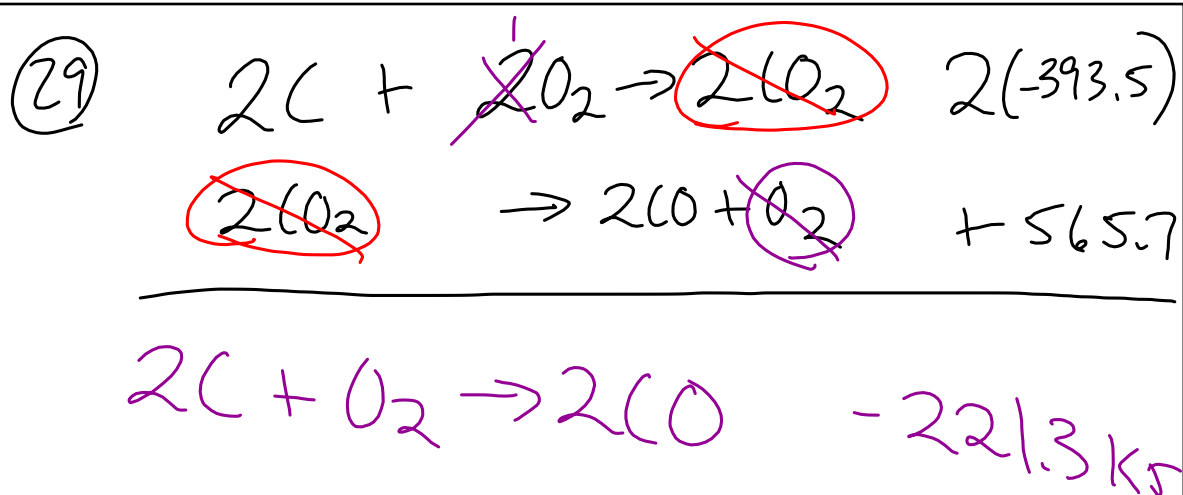
Feb 16-11:06 AM

$$\Delta G = \Delta H - T\Delta S$$

$$= -106 \text{ kJ} - (298)(0.058)$$

$$\Delta G = -123.284 \text{ kJ}$$

Feb 16-11:08 AM



Feb 16-11:09 AM

Chap 12 - GASES

Feb 16-11:14 AM