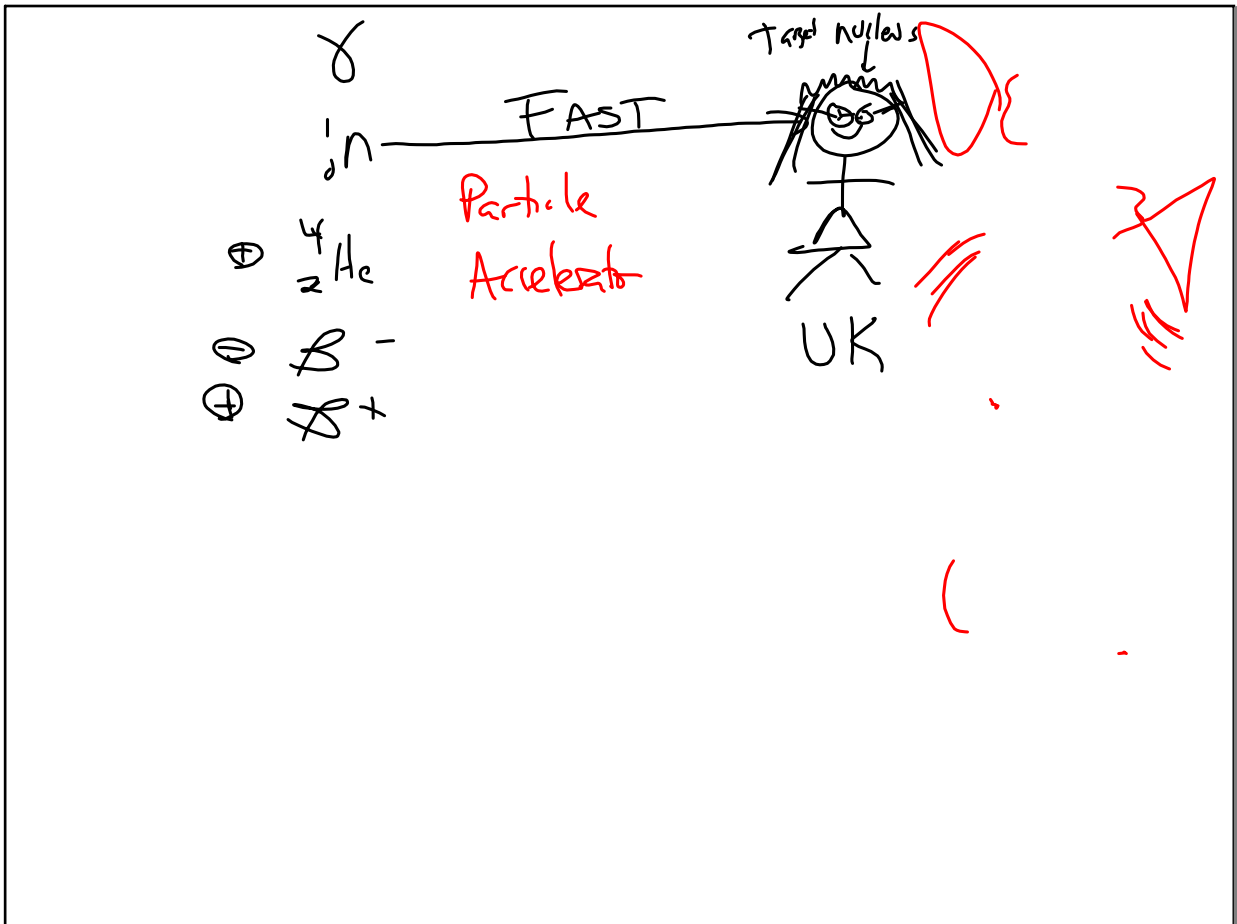
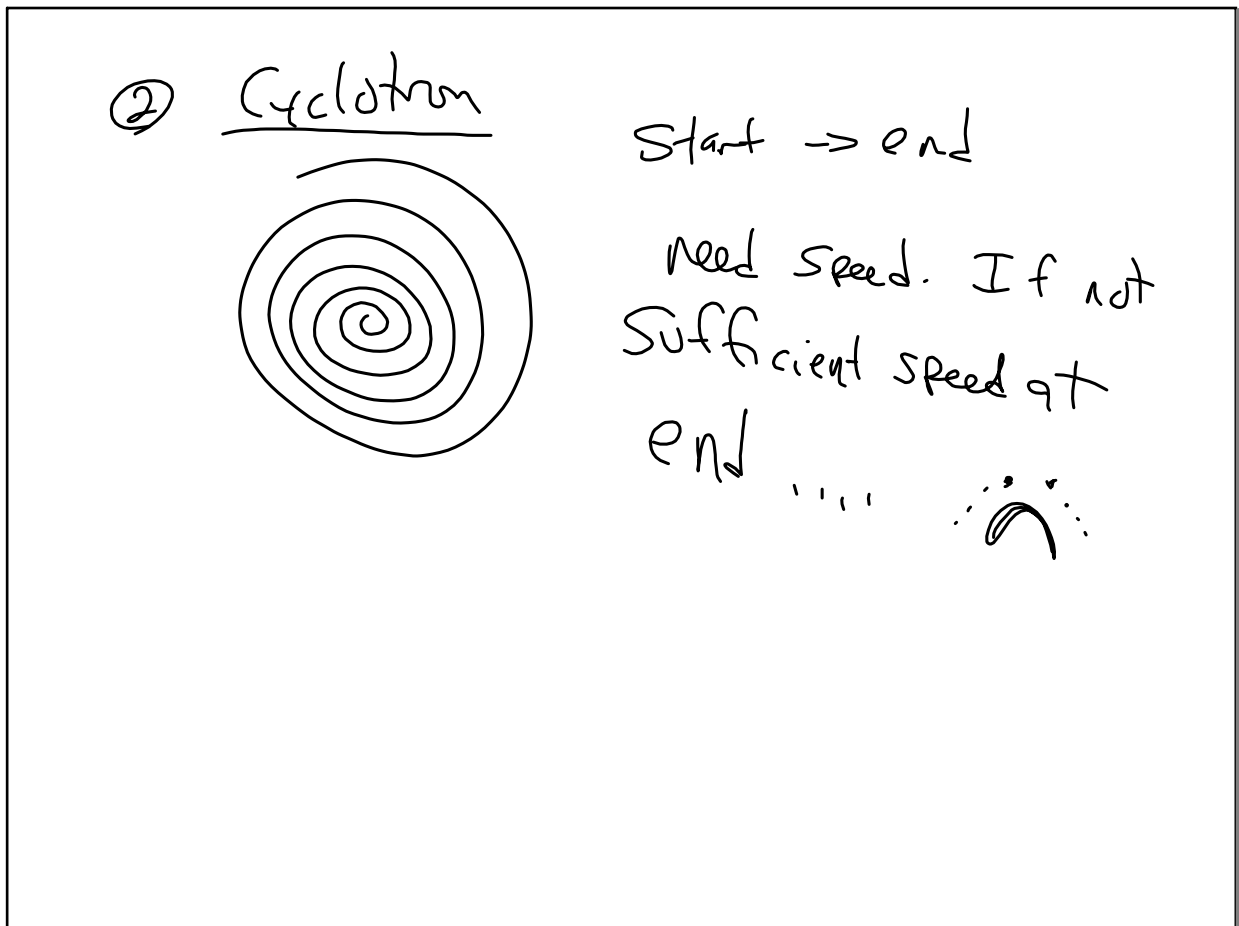
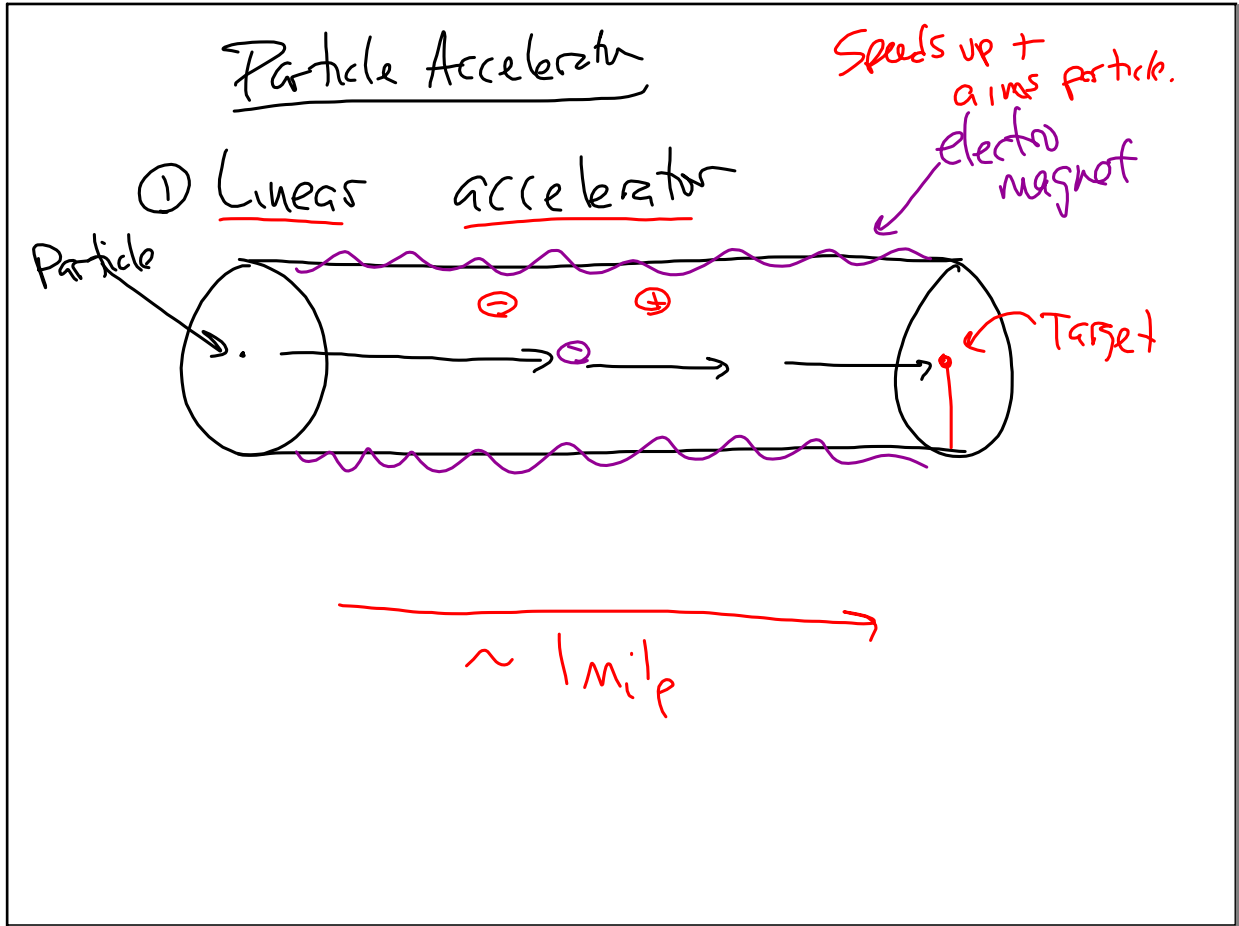


Artificial decay
Nuclear Fission

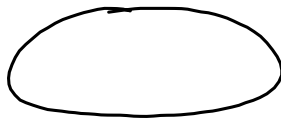
Apr 7-7:34 AM



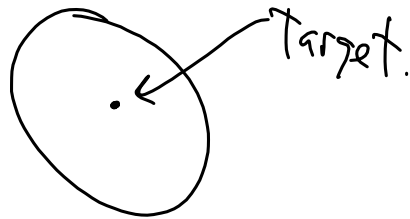
Apr 7-8:06 AM



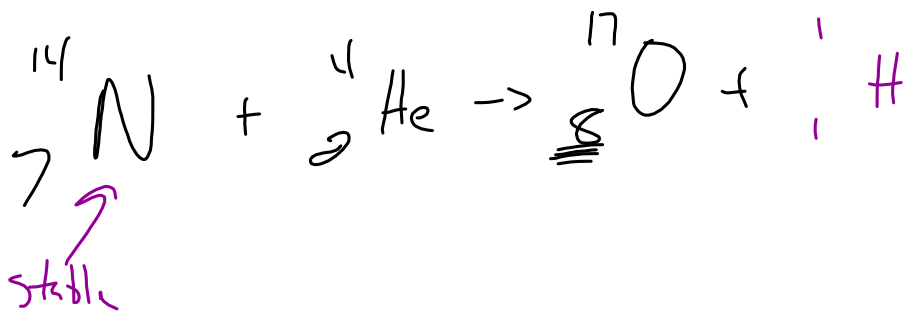
③ Synchrotron - loop



Inside



Apr 7-8:16 AM



Force to break up by SMASH with α

Apr 7-8:21 AM

Decay → calculations.

Half Life $t_{1/2}$
Time for $\frac{1}{2}$ mass to decay.

Nuclear → **FIRST ORDER RXN!**

$t_{1/2} = \frac{0.693}{k}$

$\ln A_t = -kt + \ln A_0$

How much at time "t" Rate constant Time units!

How much START

$1 \rightarrow \frac{1}{2} \rightarrow \frac{1}{4} \rightarrow \frac{1}{8} \dots$

Apr 7-8:24 AM

^{60}Co $t_{1/2} = 5.27\text{yr}$ 1mg sample

15.9 yr. mass: ^{60}Co

1mg $\xrightarrow{5.27\text{yr}}$ 0.5mg $\xrightarrow{5.27\text{yr}}$ 0.25 $\xrightarrow{5.27\text{yr}}$ 0.125mg

10.6 15.9 yr.

Apr 7-8:43 AM

③ have 1g ^{90}Sr , 0.953g remains after 2yr.

① $t_{1/2} = ?$ ② how much ^{90}Sr after 5yrs.

① $t_{1/2} = \frac{0.693}{k}$ $\ln A_t = -kt + \ln A_0$

$t_{1/2} = 28.79 \text{ yr}$ $\ln 0.953 = -k(2) + \ln 1$

$k = 0.024 \text{ yr}^{-1}$

② $\ln A_t = -kt + \ln A_0$

$\ln A_t = -(0.024)(5) + \ln 1$ $A_t = 0.886 \text{ g}$

Apr 7-8:48 AM

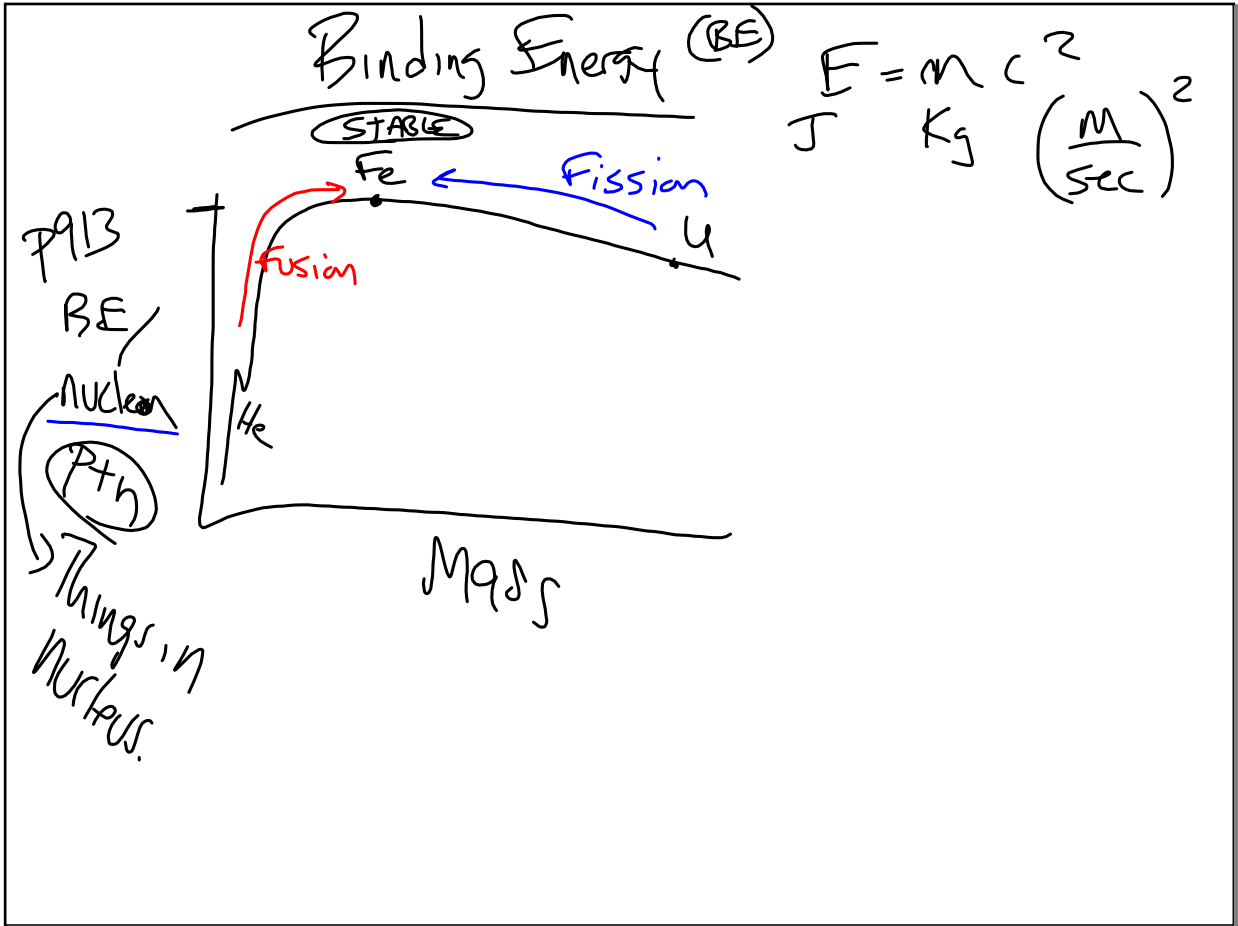
How long will it take for 60% to decompose.

$A_0 = 1$ or 100

$A_t = 0.4$ or 40

How much you HAVE

Apr 7-8:57 AM



Apr 7-9:00 AM

Measure B.F. for He

$2p = 2(1.00728)$

$2n = 2(1.00866)$

(calc)

4.03188 amu
- 4.00150
0.03038 amu

STRONG FORCE \Rightarrow

4.0015 amu

ACTUAL

Given

No es el mismo

↓

Mass Defect

↓

ENERGY to hold nucleus together

Apr 7-9:04 AM

0.03038 amu OR $\frac{g}{mole}$

$\frac{0.03038 \times 10^{-3} \text{ Kg}}{mole}$

$1 \text{ g} = 6 \times 10^{23} \text{ amu}$

$E = mc^2$

$= (5.063 \times 10^{-29}) (3 \times 10^8)^2$

$E = 4.56 \times 10^{-12} \text{ J}$

0.03038 amu	1 g	1 Kg
	$6 \times 10^{23} \text{ amu}$	1000 g

$5.063 \times 10^{-29} \text{ Kg}$

Apr 7-9:09 AM

HW

① 21/34, 36, 40

② PS 21-1 # 1-22

PS 21-2 # 1-17

Apr 7-9:14 AM