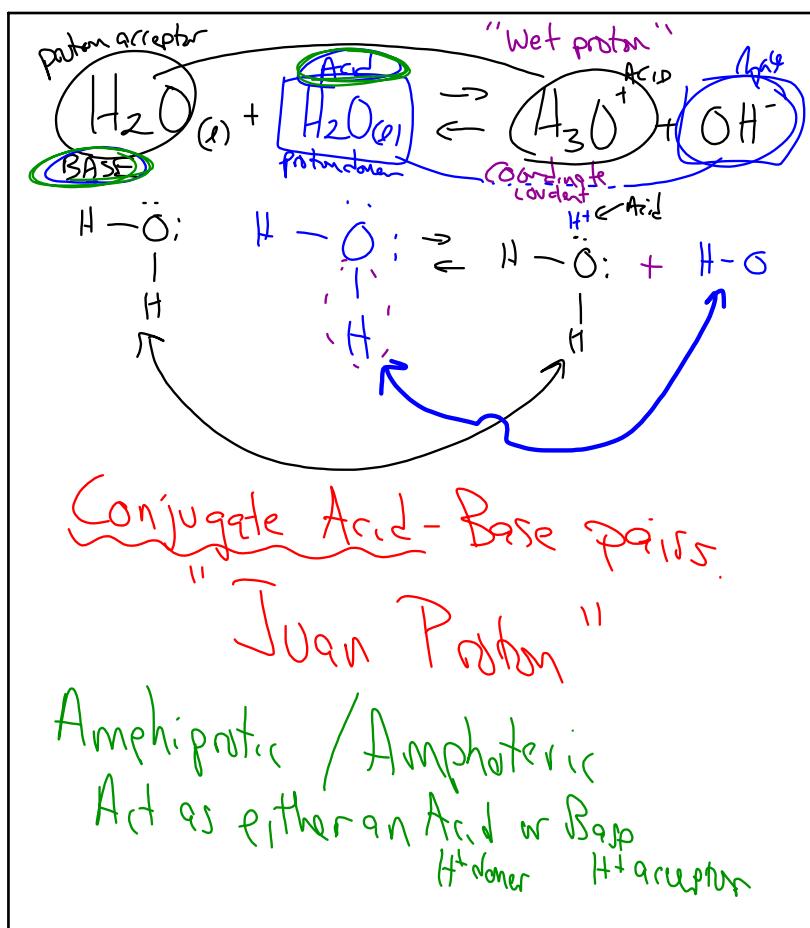




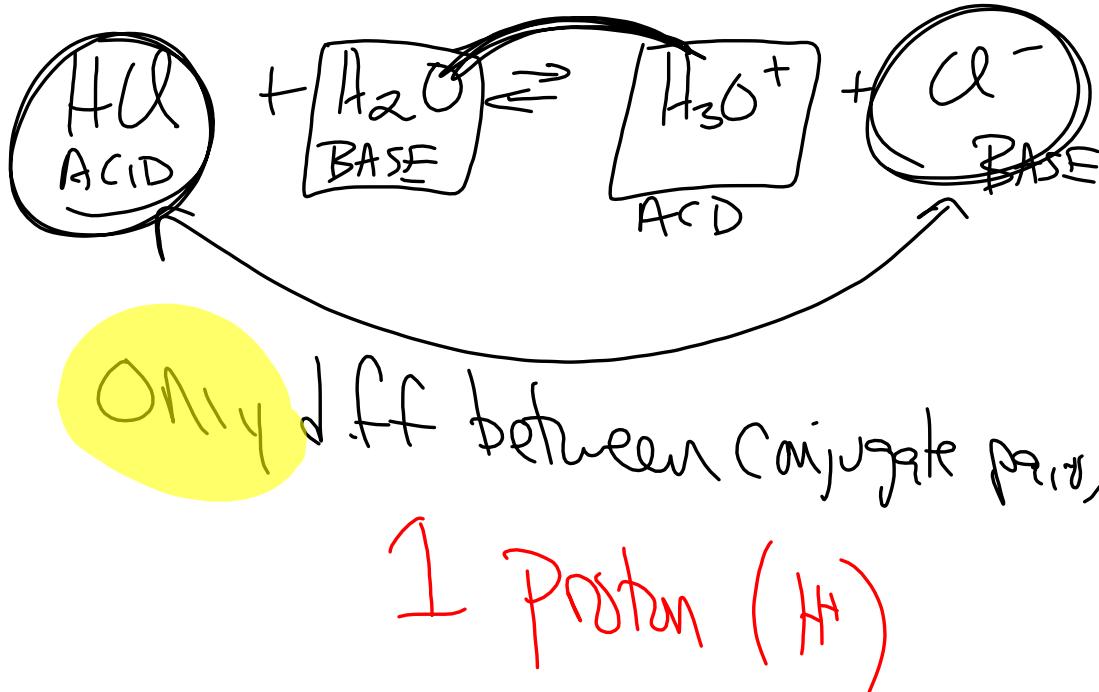
$$K_w = \frac{[\text{H}^+][\text{OH}^-]}{1} = 1 \times 10^{-14} \text{ at } 25^\circ\text{C}$$



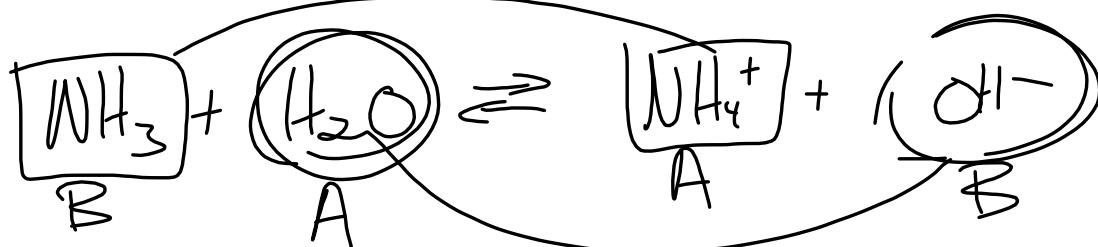
Feb 16-7:44 AM



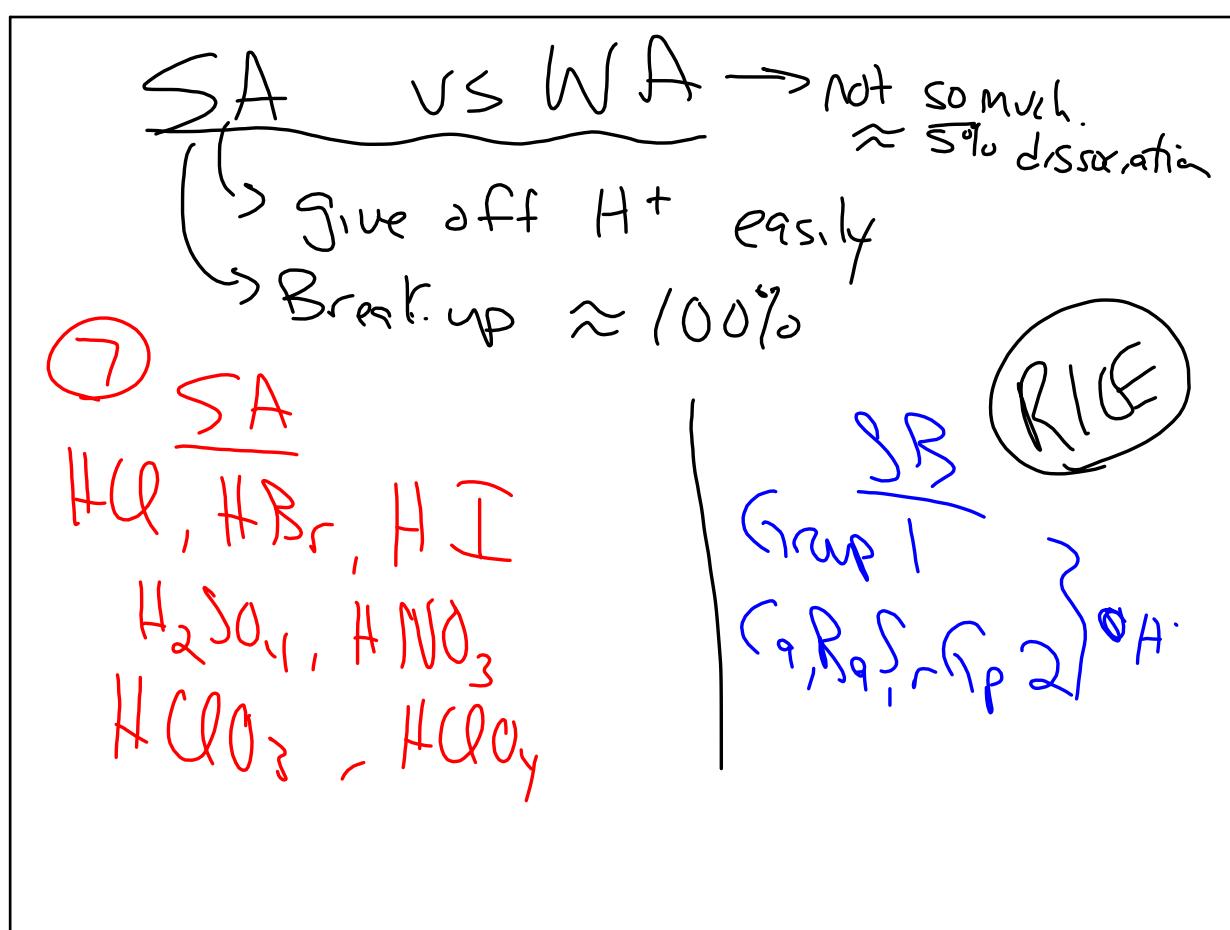
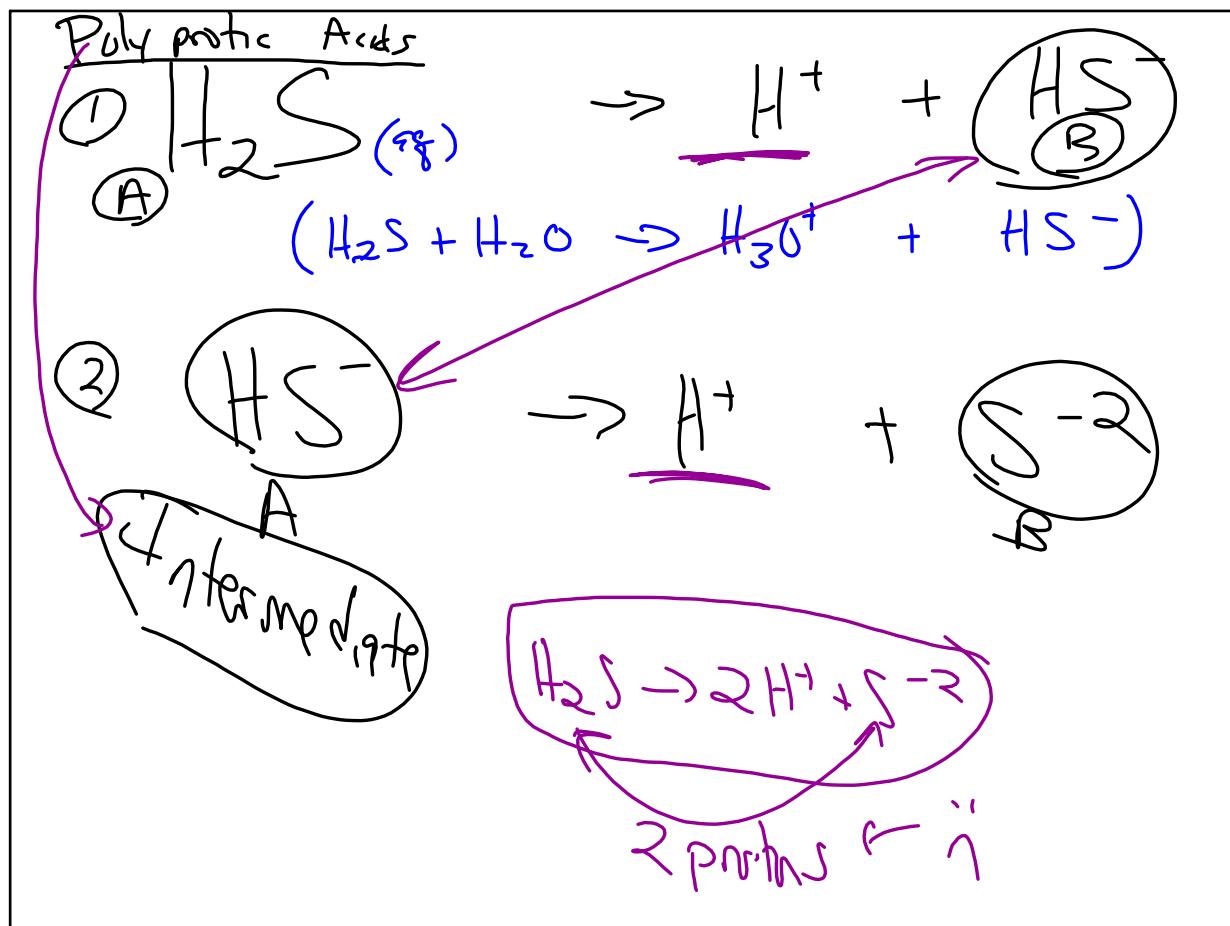
Feb 16-8:42 AM



Feb 16-8:52 AM



Feb 16-8:58 AM



$$K_w = [H^+][OH^-] = 1 \times 10^{-14} \quad @ 25^\circ C$$

$$-\log K_w = -\log(H^+)(OH^-) = 14$$

$$-\log(K_w) = -\log H^+ + -\log OH^-$$

$$pK_w = pH + pOH$$

$$pH = -\log(H^+)$$

$$pOH = -\log(OH^-)$$

$pH$ ,  $pOH$ ,  $pK_w$

Feb 16-9:07 AM

$$pH = -\log[H^+]$$

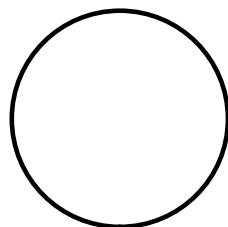
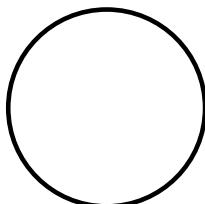
$$pOH = -\log[OH^-]$$

$$pH + pOH = 14$$

$$K_w = [H^+][OH^-] = 1 \times 10^{-14}$$

Feb 16-9:11 AM

$$pOH = 12 \quad [OH^-] = 1 \times 10^{-12}$$



Feb 16-9:12 AM

$$\checkmark H^+ = 2.73$$

$$pH = -\log(H^+) = -\log(2.73)$$

$$\checkmark pH = -0.41 \quad pOH = 14.4$$

$$[OH^-] =$$

Feb 16-9:12 AM

Before anticoagulant

$$\text{pOH} = -\log [\text{OH}^-]$$
$$14.4 = \log (\text{OH}^-)$$
$$14.4 = \log (3.98 \times 10^{-14})$$

Feb 16-9:14 AM

16 / 20, 22, 28, 38, 40

Feb 16-9:17 AM