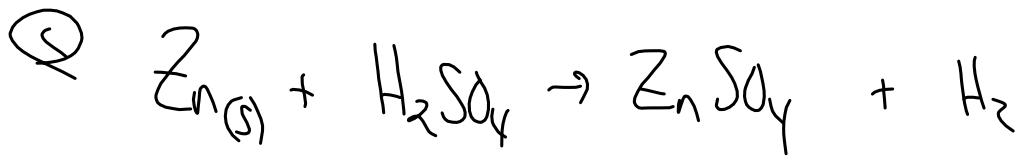
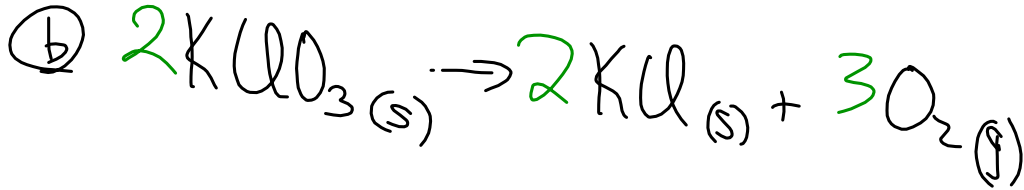
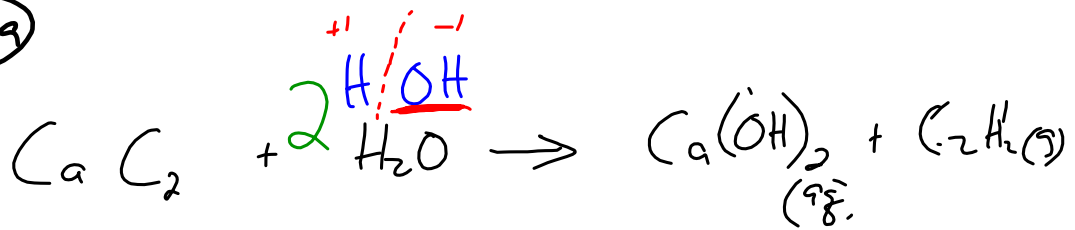


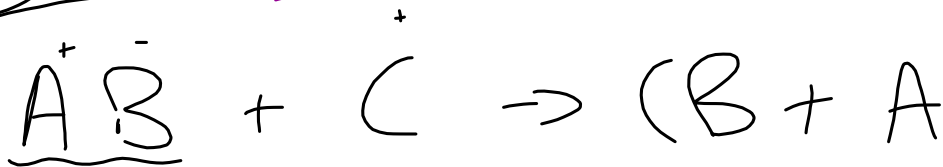
(3.133)



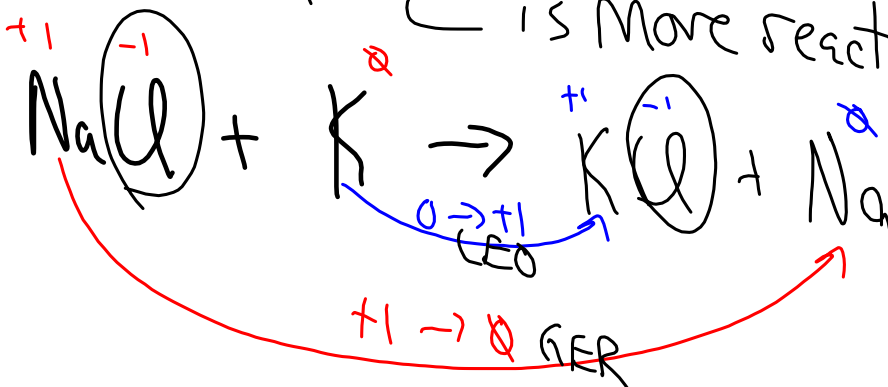
Sep 19-8:26 AM

SR

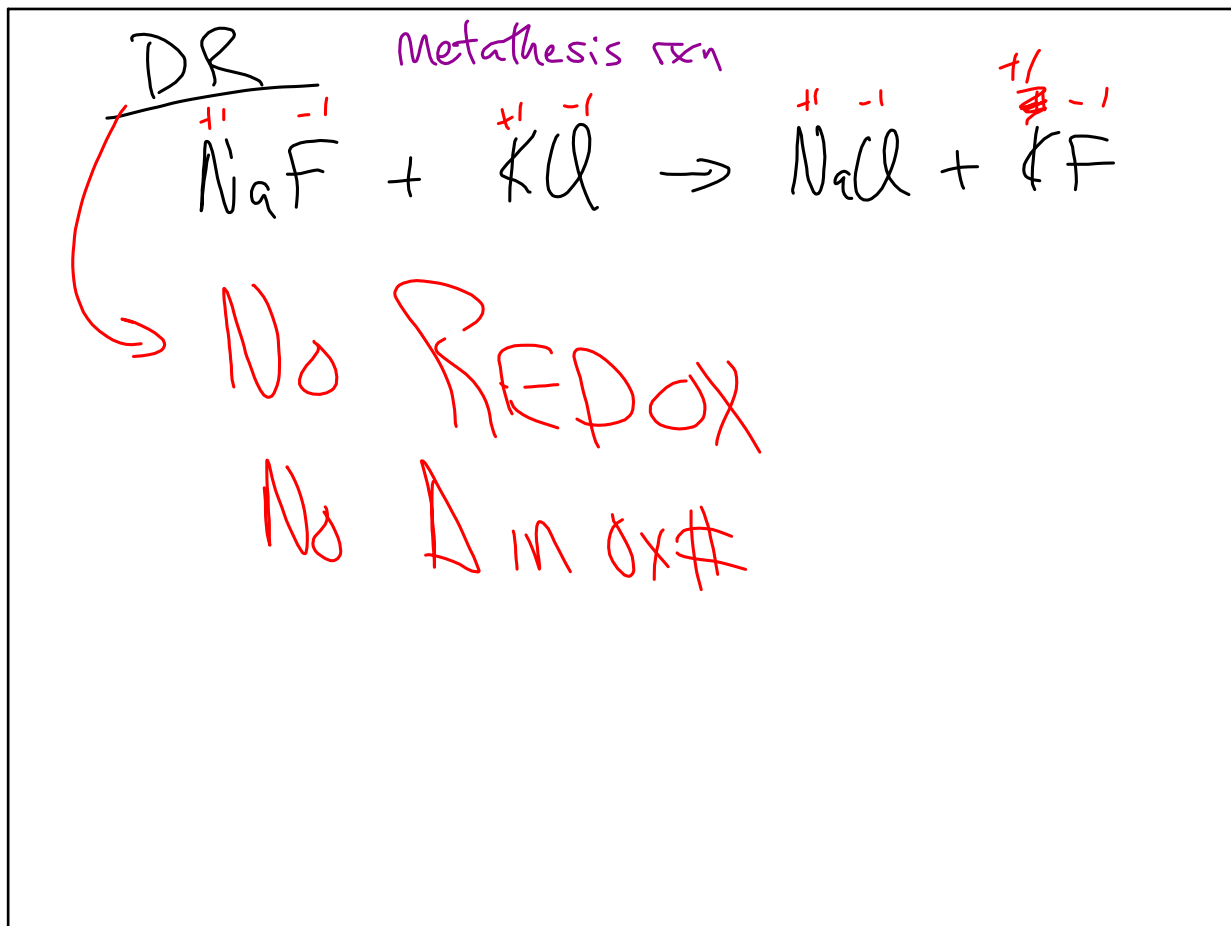
REDOX



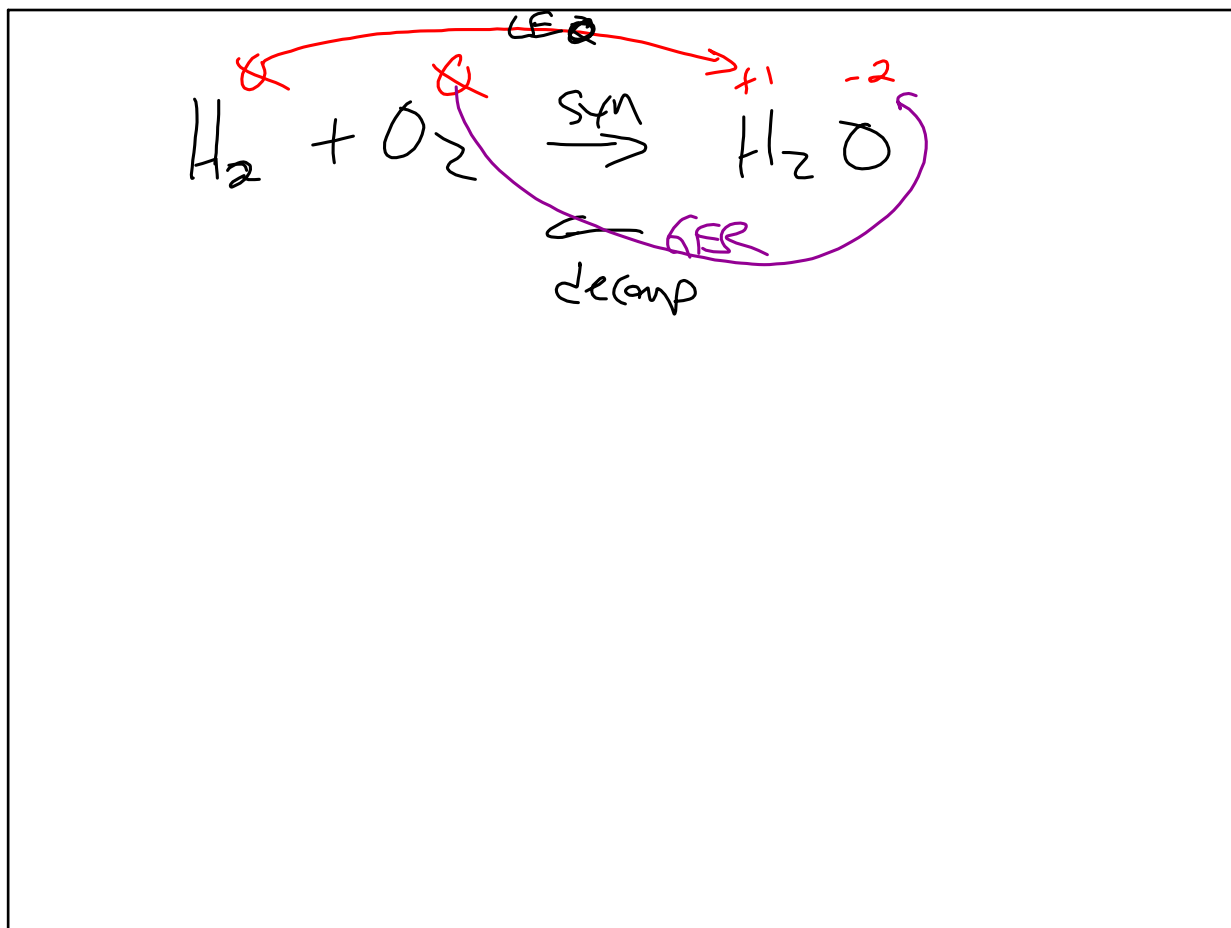
If C is more reactive than A



Sep 19-8:50 AM

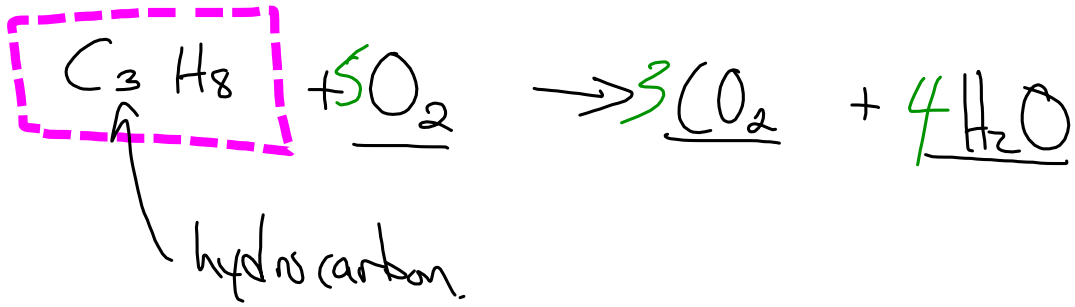


Sep 19-8:57 AM



Sep 19-8:59 AM

⑤ Combustion



50 Cars frames + 40 Tires → 10 Finished cars

- 10 used
40 left over

Used up
Limiting Reagent!

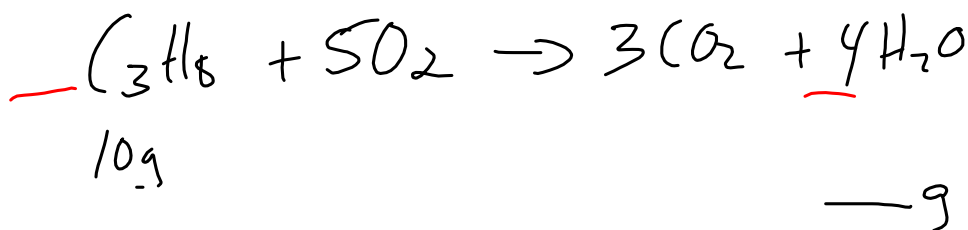
Sep 19-9:02 AM



% H in C_3H_8
by Mass

$$\frac{\text{H}_8}{\text{C}_3\text{H}_8} \times 100 = \frac{8(1)}{3(12) + 8(1)} = 18.18\%$$

Sep 19-9:11 AM



MOLE RATIOS

10g C₃H₈	1 mole C₃H₈	4 mole H₂O	18g H ₂ O
44g C ₃ H ₈	1 mole C₃H₈	1 mole C₃H₈	1 mole H₂O

Sep 19-9:13 AM

$$3 / 38 + 58 a$$

Sep 19-9:17 AM