

Observations LAB

REALLY COLD liquid !!!  
I DO NOT TOUCH!

<ul style="list-style-type: none"> <li>① Super ball (Aka grape)</li> <li>② eraser</li> <li>③ transfer pipette</li> <li>④ Army man</li> <li>⑤ Ping pong ball</li> </ul>	<ul style="list-style-type: none"> <li>⑥ Marshmallow</li> <li>⑦ rubber band</li> <li>⑧ balloon</li> </ul>	<div style="border: 1px solid blue; border-radius: 15px; padding: 5px; display: inline-block;">Macromp</div>
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Oct 2-7:28 AM

4.24g  $\overset{+3}{Cr}_2(\overset{-2}{SO}_4)_3 + 3(\overset{+1}{NH}_4)_2(\overset{-2}{CO}_3) \rightarrow \overset{+3}{Cr}_2(\overset{-2}{CO}_3)_3 + 3(\overset{+1}{NH}_4)_2\overset{-2}{SO}_4$

*molecular* →      (aq)      (aq)      (s)      (aq)

Complete ionic →      *Spectator ions → NO CHANGE.*

$$2Cr^{+3} + 3SO_4^{-2} + 6NH_4^+ + 3CO_3^{-2} \rightarrow Cr_2(CO_3)_3 + 6NH_4^+ + 3SO_4^{-2}$$

Net ionic eqn

$$2Cr^{+3} (aq) + 3CO_3^{-2} (aq) \rightarrow Cr_2(CO_3)_3 (s) \quad \checkmark$$

Oct 2-8:32 AM

(4.62)  $\ominus$  ~~11.6 M HCl~~ 0.25 mole HCl \_\_\_\_\_ ml

$\frac{11.6 \text{ moles HCl}}{1 \text{ L}} \leftarrow 1000 \text{ ml}$
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$\frac{1000 \text{ ml}}{11.6 \text{ mole HCl}}$	$\times 0.25 \text{ mole HCl} \therefore$
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Oct 2-8:44 AM