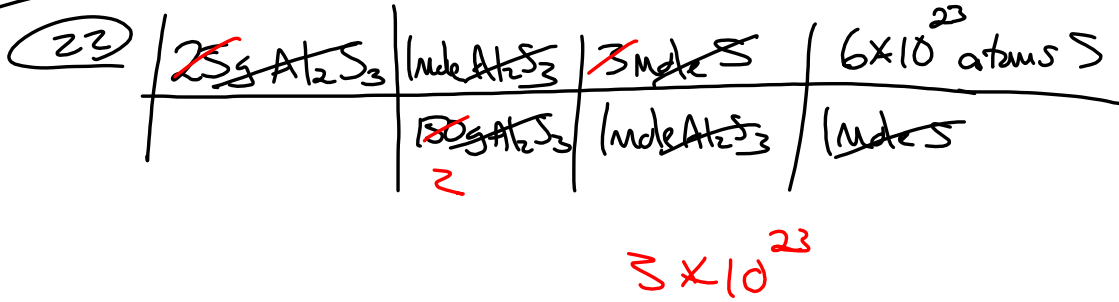


Exam 208



Oct 4-7:40 AM

(23) $\text{moles A} = \text{moles B}$

$\text{H}_2\text{SO}_4 \text{ mL} = \text{KOH mL}$

$\frac{(2)(0.737)(35)}{0.827} = \frac{(1)(0.827)}{0.827} \text{ mL}$

Oct 4-8:35 AM

$$\textcircled{27} \quad m = \frac{m \text{ols}}{l}$$

32g HCl	1 mole HCl
2.5l	36g HCl

Oct 4-8:37 AM



$$\textcircled{29} \quad \begin{array}{l} \text{Moles start} = \text{Moles end} \\ M \times l = M \times l \\ \frac{(6.75)(85.6)}{6.2 \times 1000} = \frac{M (6.2 \times 1000)}{6.2 \times 1000} \end{array}$$

Oct 4-8:38 AM